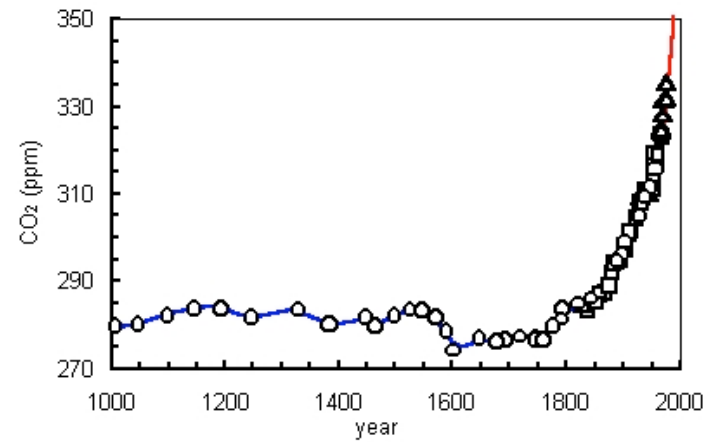


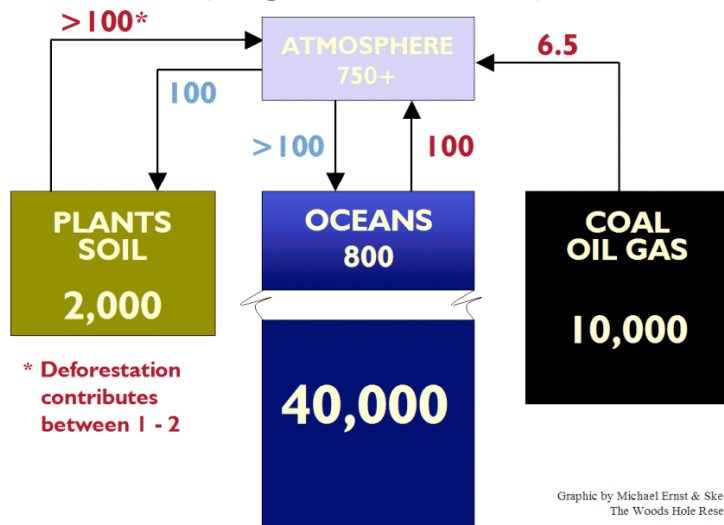


The Carbon Cycle

The carbon dioxide record in the Greenland Ice Core



Global Flows of Carbon
(Petagrams of Carbon/Year)



There are three main types of fossil fuels:

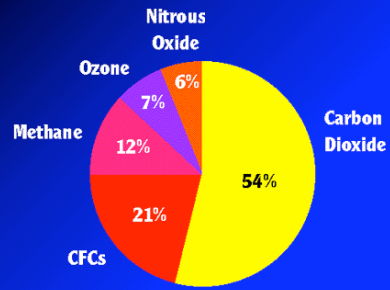
- (1) Oil and its derivatives
- (2) Natural Gas
- (3) Coal

Fossil fuels are typically composed of compounds containing hydrogen and carbon atoms. The process of burning fossil fuels always results in carbon dioxide being emitted into the atmosphere. Take the example of natural gas (methane).



Water and carbon dioxide are the by-products

Contributions to Global Warming by the Major Greenhouse Gases



Source - The Greenhouse Trap, Francesca Lynson, World Resources Institute, 1990

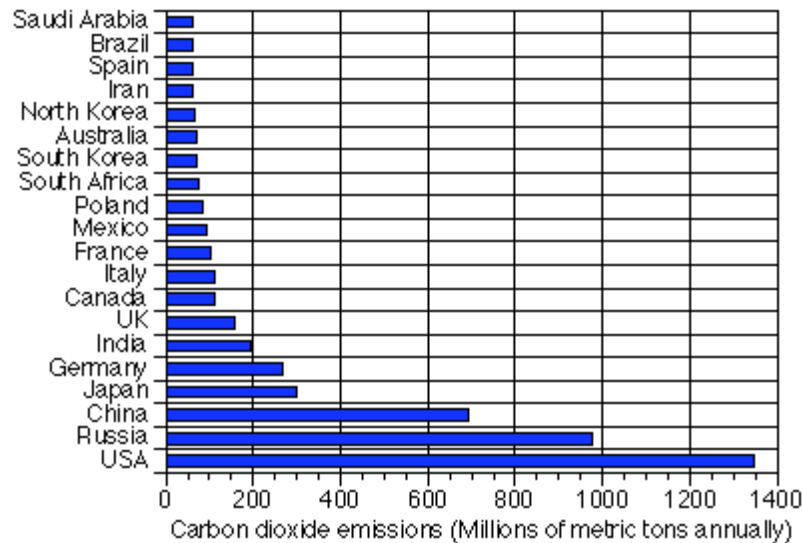
CG Figure-22



It is critical to understand what controls the CO₂ concentrations in the atmosphere, because the increase in CO₂ makes up more than half of the human enhancement of the greenhouse effect. This is the justification for studying the global carbon cycle.

Let's begin our examination of the carbon cycle by focusing on emissions of CO₂. Currently, about 6.3 gigatons of carbon are released into the atmosphere as a result of fossil fuel burning.

About 40% of this comes from coal burning, another 40% comes from burning of oil and oil derivatives (such as gasoline), and the remaining 20% comes from burning of natural gas (methane).



Here's how the total emissions break down by country.

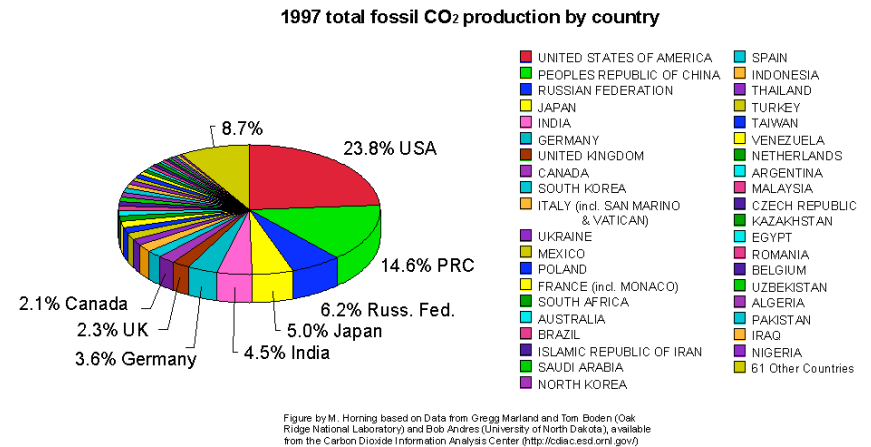
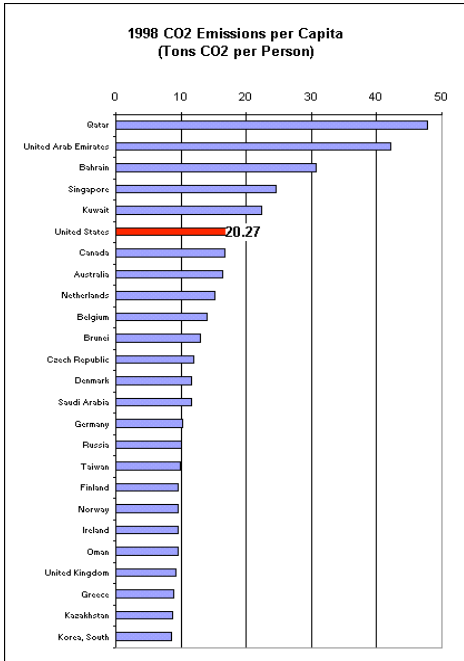
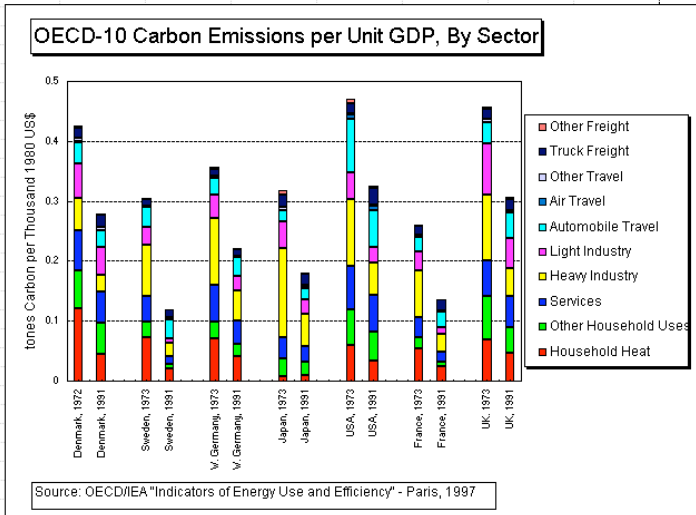
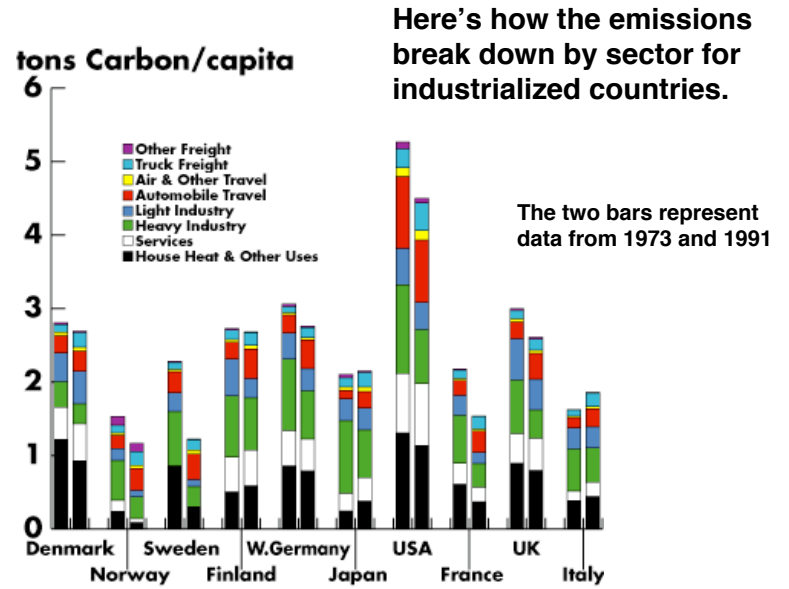


Figure by M. Horning based on Data from Gregg Marland and Tom Boden (Oak Ridge National Laboratory) and Bob Andres (University of North Dakota), available from the Carbon Dioxide Information Analysis Center (<http://cdiac.esd.ornl.gov/>)

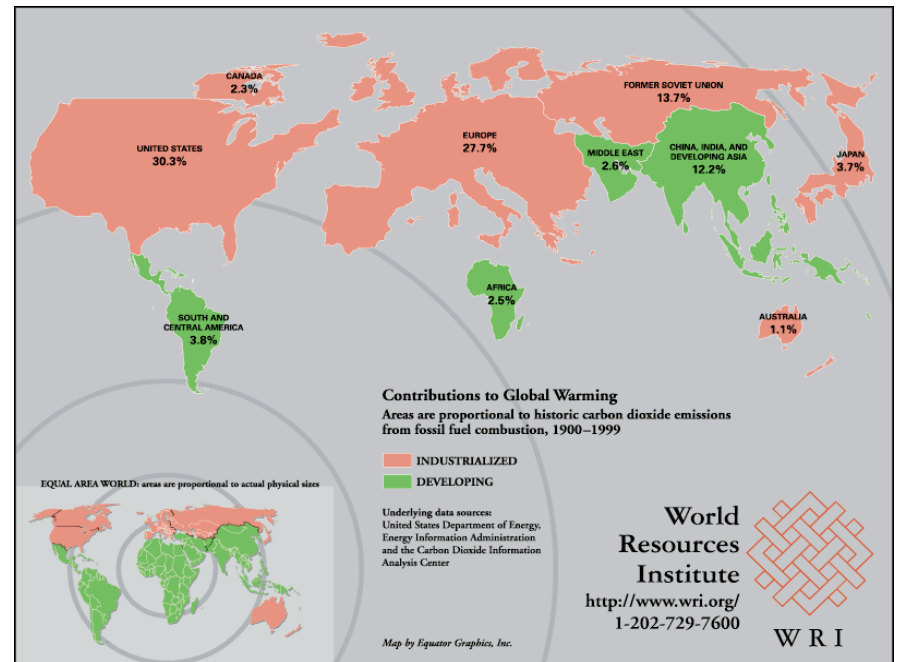
Here's how each country's emissions break down as a percentage of the total.



We've seen that the U.S. produces by far the most emissions of CO2 of any country in the world. How does the U.S. compare to other countries on a per capita basis?

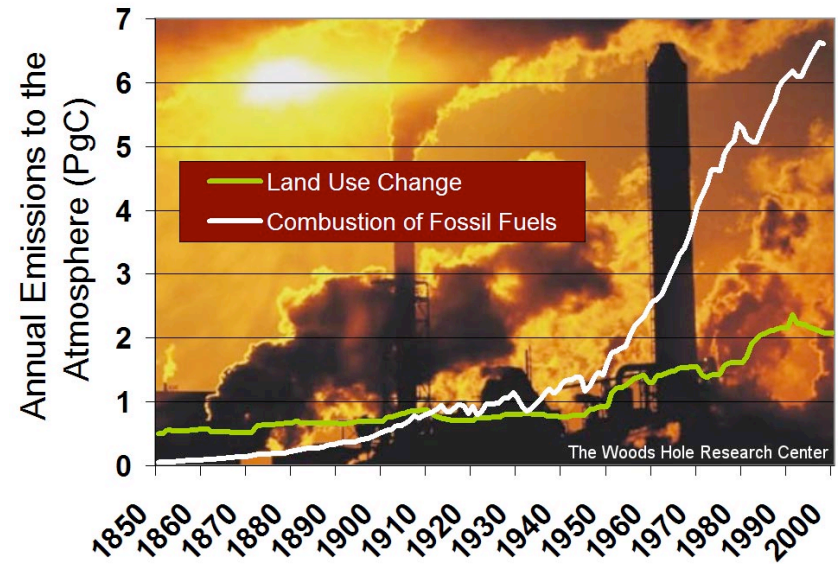


The U.S. does a little better when emissions are normalized by economic output.

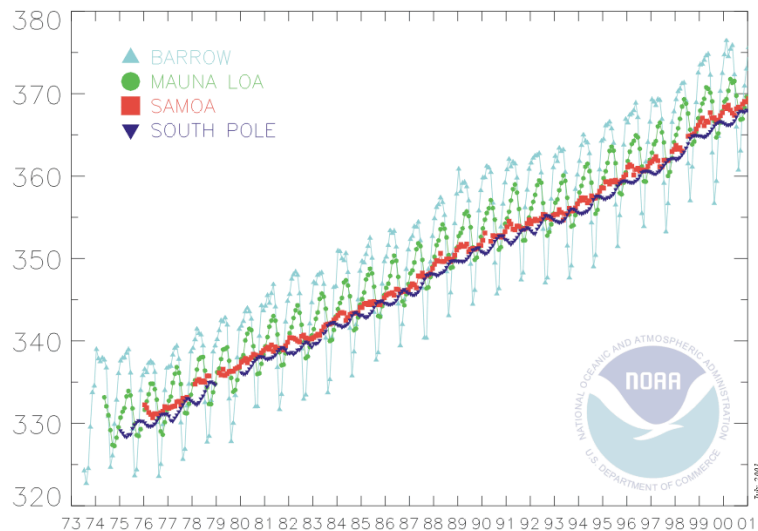


Human activity is also adding CO₂ to the atmosphere through deforestation. When tropical forests are clear cut, the land is typically converted to pasture. The original forest and its soil have a much higher carbon content than the pastureland, so the process of burning the forest must result in a significant release of carbon to the atmosphere.

Deforestation is thought to be releasing about 2.2 gigatons of carbon every year into the atmosphere.



Carbon Dioxide Measurements



Now let's consider what happens to all that CO₂ that is emitted into the atmosphere.

Of the 6.3 gigatons of carbon being put into the atmosphere every year by fossil fuel burning and the 2.2 gigatons being put there by deforestation (total 8.5 gigatons), only about 3.2 gigatons actually remains there.

The amount remaining can be calculated directly from the increase in atmospheric CO₂.

So where is all the excess CO₂ going?

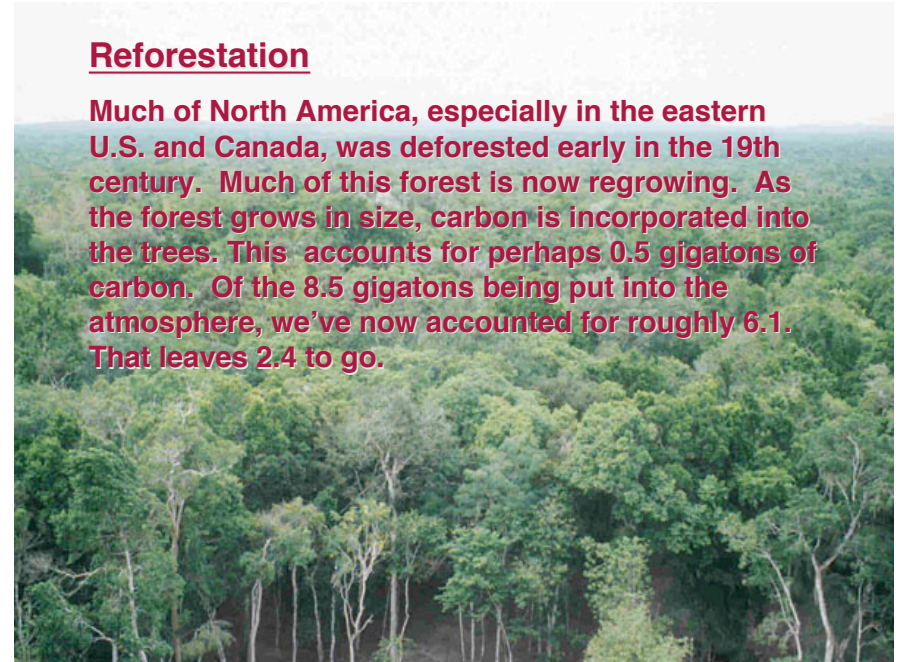
Uptake by the Ocean

The main way anthropogenic CO₂ is removed from the atmosphere is through oceanic uptake. This occurs because CO₂ dissolves in seawater. The excess carbon is eventually incorporated into the skeletons of marine organisms and buried in deep sea sediments as the organisms die and fall to the bottom of the ocean.

This process is thought to remove about 2.4 gigatons of carbon per year from the atmosphere. So we started with 8.5 gigatons, 3.2 are in the atmosphere, 2.4 are taken up by the ocean, leaving us with 2.9 gigatons still to be accounted for.

Reforestation

Much of North America, especially in the eastern U.S. and Canada, was deforested early in the 19th century. Much of this forest is now regrowing. As the forest grows in size, carbon is incorporated into the trees. This accounts for perhaps 0.5 gigatons of carbon. Of the 8.5 gigatons being put into the atmosphere, we've now accounted for roughly 6.1. That leaves 2.4 to go.



CO₂ fertilization

Plants need CO₂ for photosynthesis. They generally obtain this CO₂ through **stomata**, small openings on their leaves. But plants also lose water, another critical substance for their survival, through their stomata. This makes it essential that they be able to control the size of the stomata. When the CO₂ concentration is increased, the stomata do not have to be as large to take in the same amount of CO₂. So the plants can survive and continue to grow under drier conditions.

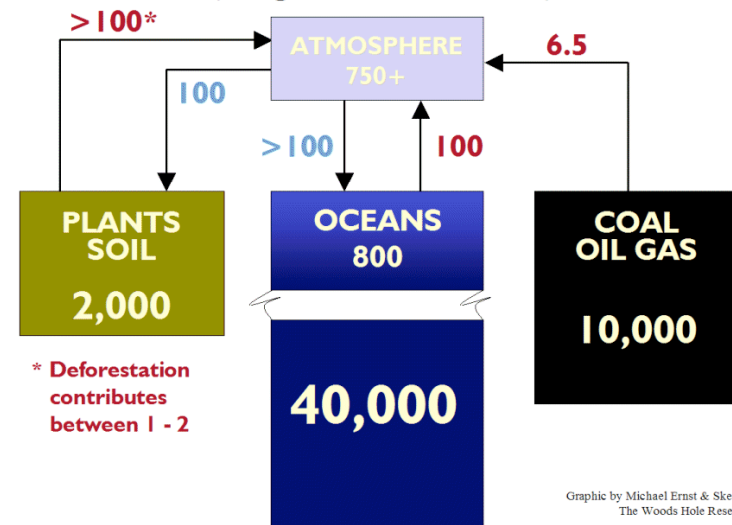
It is possible that increased CO₂ levels in the atmosphere are leading to enhanced growth rates of terrestrial plants, accounting for some of the missing carbon.



Microscopic view of the underside of a leaf surface

Global Flows of Carbon

(Petagrams of Carbon/Year)



Graphic by Michael Ernst & Skee Houghton
The Woods Hole Research Center